# TWO MINOR DITERPENES RELATED TO DICTYODIAL A FROM THE BROWN ALGA DICTYOTA CRENULATA

## MICHAEL P. KIRKUP and RICHARD E. MOORE

Department of Chemistry, University of Hawaii, Honolulu, HI 96822, U.S.A.

(Received 12 February 1983)

Key Word Index—Dictyota crenulata; Dictyotaceae; brown alga; diterpenes; structure determination.

Abstract— $4\beta$ -Hydroxydictyodial A and 18,0-dihydro- $4\beta$ -hydroxydictyodial A 18-acetate are two minor aldehydic diterpenes in the brown alga *Dictyota crenulata*. <sup>1</sup>H NMR and lanthanide-induced chemical shift studies indicate that the relative stereochemistry of both compounds is  $2R^*, 3R^*, 4S^*, 10S^*$ .

#### INTRODUCTION

Dictyodial A (1) is the major diterpene in the lipophilic extract of the marine brown alga Dictyota crenulata (Kutzig) from Hawaii [1]. During the winter months this seaweed produces minor amounts of a novel aldehydic diterpene,  $\beta$ -crenulal [2], along with two diterpenes that are related to dictyodial A. We report here the structures and relative stereochemistries of the two dictyodial A-related diterpenes.

## RESULTS AND DISCUSSION

One diterpene has been identified as  $4\beta$ -hydroxydictyodial A (2). Mass spectrometry indicates that the molecular formula is C<sub>20</sub>H<sub>30</sub>O<sub>3</sub> and the IR spectrum shows that a hydroxyl group is present (3440 cm<sup>-1</sup>). The <sup>13</sup>C NMR and <sup>1</sup>H NMR spectra of 2 (Table 1) closely resemble those of dictyodial A [1]. Signals are present for two aldehydic functionalities, one of which is in a conjugated enal system ( $\lambda_{max}$  231 nm); the chemical shift of the proton (H-18) in the unconjugated aldehyde group of 2, however, is 0.5 ppm higher field than that of 1, suggesting that the OH group is close to H-18. Detailed analysis (spin-spin decoupling) has established that the OH group is attached  $\beta$  to C-4, since the proton on C-4  $(\delta 4.32)$  shows small to nil couplings (0-4 Hz) to all of its vicinal neighbors. Inspection of a Dreiding model of 2 (in the conformation suggested from X-ray crystallographic analysis of dictyodiol and dictyolactone, two derivatives of 1[1]) shows that when the OH on C-4 is  $\beta$ , the dihedral angles between H-4 and H-3, H-4 and H-5 $\beta$ , and H-4 and H-5α are in the 30-90° range, in perfect agreement for the 0-4 Hz couplings that are seen between these pairs of protons. Zero coupling is seen between H-2 and H-3, indicating that the dihedral angle between these two protons is ca 90°, as it is in 1. H-3 resonates as a broad singlet at  $\delta$  1.60, showing nil coupling to not only H-2 and H-4, but to H-10 as well  $(J_{3,10} \text{ for 1 is } \sim 0.5 \text{ Hz}[1])$ . The relative stereochemistry of 2, therefore, appears to be the same as 1 at C-2, C-3 and C-10. Since the optical rotations are almost the same for  $2([\alpha]_D - 121^\circ)$  and  $1([\alpha]_D - 95^\circ)$ and since the <sup>1</sup>H NMR data strongly suggest that both diterpenes have virtually the same conformation in solution, the absolute stereochemistry of 2 is probably the same as that of 1 (the absolute stereochemistry of 1, however, is presently unknown).

The second diterpene has structure 3. Mass spectrometry establishes its elemental composition as C<sub>22</sub>H<sub>34</sub>O<sub>4</sub>. The presence of OH and acetoxyl groups are indicated by

- 1  $R^1 = CHO, R^2 = H$
- 2  $R^1 = CHO, R^2 = OH$
- 3  $R^1 = CH_2OAc_1R^2 = OH$

- 4 R = H
- 5 R = OH

Table 1. NMR spectral data for  $4\beta$ -hydroxydictyodial A (2) in CDCl<sub>3</sub>

<sup>13</sup> C δ*	Assignment	$^{1}$ H $\delta$ †
203.6 (d)	18	9.69 (1H, s)
194.2 (d)	19	9.39 (1H, s)
157.8 (d)	9	7.06 (1H, dd)
147.9 (s)	1	
138.1 (s)	6	
131.2 (s)	14	
124.6 (d)	13	5.00 (1H, tm)
124.2 (d)	7	5.33 (1H, ddm)
73.6 (d)	4	4.32 (1H, br m)
52.0 (t)	8 <sub>\$\beta\$</sub>	3.43 (1H, ddd)
	8,	3.10 (1H, ddd)
50.5 (d)	3	1.60 (1H, br s)
47.9(t)	5 <sub>2</sub>	2.46 (1H, dd)
	5 <sub>\$\beta\$</sub>	2.09 (1H, dd)
39.59(t)	11	$1.10 \ (2H, br \ m)$
33.25 (d)	10	2.11 (1H, obscured m)
29.55 (d)	2	3.58 (1H, br s)
25.68 (t)	12	1.82 (2H, m)
25.68 (q)	16	1.58 (3H, br m)
20.22 (q)	17	1.13 (3H, d)
17.58 $(q)$	15	1.68 (3H, $br m$ )
17.58 (q)	20	1.98 (3H, br m)

\*25.2 MHz, TMS as internal standard = 0. †220 and 100 MHz, TMS as internal standard = 0. J(Hz): 2, 3 = 0; 3, 4 = 1; 3, 5<sub>x</sub> = 1; 4, 5<sub>p</sub> = 1; 4, 5<sub>z</sub> = 2.5; 5<sub>x</sub>, 5<sub>p</sub> = -13; 7, 8<sub>p</sub> = 11; 7, 8<sub>x</sub> = 3.5; 7, 20 = 1.5; 8<sub>x</sub>, 8<sub>p</sub> = -16; 8<sub>p</sub>, 9 = 3.5; 8<sub>x</sub>, 9 = 8; 10, 17 = 7; 11, 12 = 7; 12, 13 = 7; 13, 15 = 1.5; 13, 16 = 1.5.

the mass spectrum since fragment ions are observed for successive losses of H2O and HOAc from the molecular ion. The IR spectrum provides further evidence for the presence of these OH  $(3500\,\mathrm{cm^{-1}})$  and acetoxyl  $(1730\,\mathrm{cm^{-1}})$  functionalities. The UV spectrum shows that an enal group is present ( $\lambda_{max}$  244 nm). The <sup>1</sup>H NMR spectrum of 3 (Table 2) is very similar to that of 2, except that acetoxymethyl signals are found instead of the signal for the unconjugated aldehyde group. The coupling constants for the various ring protons in 3 are identical with those in 2. Lanthanide-induced chemical shift studies of 3 (Table 2) prove that the OH on C-4 is  $\beta$ . H-18, H-18 and H-2 shift very rapidly, in fact faster than H-4, and much faster than the protons on C-7, C-9, C-19 and C-20, when Eu(fod)3 is added to the NMR sample, establishing that the OH on C-4 is very close to H-2, H-18 and H-18'. A Dreiding model shows that this is possible only when 3 has the same conformation indicated by X-ray analysis for dictyodiol and dictyolactone [1] and when the OH on C-4 is  $\beta$ . Diterpene 3 also has a laevorotatory optical rotation  $([\alpha]_D - 187^\circ)$ , suggesting that its absolute stereochemistry is probably the same as that of 1 and 2.

The relative stereochemistry of both 2 and 3 is apparently  $2R^*$ ,  $3R^*$ ,  $4S^*$ ,  $10S^*$ .

 $\beta$ -Crenulal (4) is readily formed when dictyodial A is treated with boron trifluoride etherate in benzene at 5° [2]. Attempted cyclization of 2 to 5 under the same conditions leads to anomalous products.

Table 2. <sup>1</sup>H NMR spectral and lanthanide-induced chemical shift data for 3

	δ*	Δδ†		δ*	<b>Δ</b> δ†
H-19	9.32 d	2.80	H-5 <sub>a</sub>	br d	
H-9	6.84 dd	1.73	H-5 <sub>8</sub>	dd	
H-7	5.25 ddq	2.18	AcÓ	S	
H-13	5.04 t septets		3H-20	br m	2.70
H-18	4.65 dd	7.30	2H-12	br m	
H-18'	4.54 dd	7.95	3H-15	br s	
H-4	4.29 m	5.63	3H-16	br s	
H-8 <sub>8</sub>	3.34 ddd		H-3	br m	
H-8,	3.01 ddd		2H-11	br m	
H-2	3.24 ddd	6.50	3H-17	d	
H-10	2.32 m				

<sup>\*220</sup> MHz, CDCl<sub>3</sub>, TMS = 0 as internal reference.

J(Hz): 2, 3 = 0; 2, 18 = 7; 2, 18' = 9; 2, 19 = 2; 3.4 ~ 0; 3, 10 ~ 0; 4,  $5_x$  = 2; 4,  $5_\beta$  = 4;  $5_x$ ,  $5_\beta$  = -13; 7,  $8_x$  = 3.5; 7,  $8_\beta$  = 11; 7, 20 = 1.5;  $8_x$ ,  $8_\beta$  = -15;  $8_x$ , 9 = 8;  $8_\beta$ , 9 = 3.5; 10, 17 = 7; 11, 12 = 7.5; 12, 13 = 7.5; 12, 15 = 1.5; 12, 16 = 1.5; 13, 15 = 1.5; 13, 16 = 1.5; 18, 18' = -12.

## EXPERIMENTAL

Isolation. Wet Dictyota crenulata (Kutzig) (397 g dry wt) collected at Kualoa Beach Park, Oahu, Hawaii in January 1977 was extracted with Me<sub>2</sub>CO and CH<sub>2</sub>Cl<sub>2</sub>. The dried extract was distributed between H<sub>2</sub>O and CH<sub>2</sub>Cl<sub>2</sub> and the organic phase was evapd to 35.6 g of a dark green oil which was chromatographed on a 70 cm × 6.5 cm column of silica gel with hexane, hexane-CHCl<sub>3</sub> and CHCl3. The fraction eluted with CHCl3 (4.71 g) was then subjected to gel permeation chromatography in 500 mg portions on a 120 cm × 3 cm column of Biobeads SX-8, a porous styrene-divinylbenzene copolymer, with C<sub>6</sub>H<sub>6</sub>. HPLC of the material eluted after the pigments on a Whatman Partisil 10/50 M-9 column with 5 % EtOAc-CHCl<sub>3</sub> gave 278 mg of  $\beta$ -crenulal, 100 mg of  $4\beta$ -hydroxydictyodial A (2, 0.28%) and 125 mg of 18,0-dihydro-4 $\beta$ -hydroxydictyodial A 18-acetate (3, 0.35%). Purification of 2 and 3 was also achieved by reverse-phase HPLC on a Whatman Partisil 10/50 ODS-2 column with 15% H<sub>2</sub>O-MeCN.

 $4\beta$ -Hydroxydictyodial A (2) was isolated as an oil,  $[\alpha]_D^{25} = 121^\circ$  (c 0.33; EtOH); IR  $v_{max}^{neal}$  cm<sup>-1</sup>: 3440 (OH), 2940, 2890, 2710, 1695 (sh), 1670, 1610, 1440, 1380, 1180, 1110, 840; UV  $\lambda_{max}^{EtOH}$  nm (ε): 232 (4000); EIMS (probe) 70 eV, m/z (rel. int.): 318.219 [M]<sup>+</sup> (4) C<sub>20</sub>H<sub>30</sub>O<sub>3</sub>, 300 [M - H<sub>2</sub>O]<sup>+</sup> (12), 262 (7), 261 (7), 201 (8), 189 (11), 177 (8), 176 (7), 165 (11), 161 (26), 147 (15), 135 (19), 109 (38), 95 (25), 93 (25), 91 (23), 82 (22), 69 [Me<sub>2</sub>C=CH-CH<sub>2</sub>]<sup>+</sup> (100), 67 (23), 45 (18), 43 (15).

18,O-Dihydro-4β-hydroxydictyodial A 18-acetate (3) was isolated as a colorless oil,  $[\alpha]_{25}^{25} - 187^{\circ}$  (c 0.80; EtOH); IR  $v_{\rm max}^{\rm neat}$  cm<sup>-1</sup>: 3500, 2940, 2900, 2840, 2700, 1735, 1690, 1445, 1370, 1240, 1110, 1030, 800; UV  $\lambda_{\rm max}^{\rm EtOH}$  nm (ε): 244 (2500), 270 (sh. 1560); EIMS (probe) 70 eV, m/z (rel. int.): 362.217 [M]<sup>+</sup> (6)  $C_{22}H_{34}O_4$ , 344 [M – H<sub>2</sub>O]<sup>+</sup> (5), 318 (5), 302 [M–HOAc]<sup>+</sup> (16), 284 [M – H<sub>2</sub>O – HOAc]<sup>+</sup> (14), 149 (46), 109 (47), 107 (30), 105 (34), 95 (38), 93 (34), 91 (38), 81 (68), 69 [Me<sub>2</sub>C=CH–CH<sub>2</sub>]<sup>+</sup> (100), 55 (48), 43 (56), 41 (70).

Acknowledgements—This research was supported by Grants CHE76-82517 and CHE79-25416 from the National Science Foundation. The 220 MHz NMR studies were carried out at the

<sup>†100</sup> MHz, CDCl<sub>3</sub>, 1 molar equivalent Eu(fod)<sub>3</sub>.

University of California, San Diego.

## REFERENCES

1. Finer, J., Clardy, J., Fenical, W., Minale, L., Riccio, R., Battaile,

 J., Kirkup, M. and Moore, R. E. (1979) J. Org. Chem. 44, 2044.
Kirkup, M. P. and Moore, R. E. (1978) 175th National Meeting of the American Chemical Society, Anaheim, California, March 1978, Abstract No. ORGN 187.